

Neutral Compound Practice

To write a neutral compound, all charges must _____

Ionic compounds are written with the _____ first, and the _____ second.

	Iodide	Fluoride	Oxide	Phosphide	Nitrate	Sulfate
Calcium						
Silver						
Barium						
Lead (II)						
Chromium(III)						
Tin (IV)						
Ammonium						
Copper (II)						
Iron (II)						
Aluminum						
Magnesium						
Lithium						
Iron (III)						
Vanadium(V)						